

CORRELATIONS IN Cu- AND Mn-BEARING TOURMALINES FROM BRAZIL AND MOZAMBIQUE

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Cu- and Mn-bearing tourmalines from Brazil and Mozambique were characterised chemically (EMPA and LA-ICP-MS) and by single-crystal structure refinement. All these samples are rich in Al, Li and F (fluor-elbaite) and contain significant amounts of CuO (up to ~1.8 wt%) and MnO (up to ~3.5 wt%). MgO and FeO contents are relatively low in the 15 investigated samples (≤ 0.2 wt%). Single-crystal structure refinements were done on 8 samples.

Tourmaline is a silicate mineral group with a highly complex crystal structure and a large variety of chemical compositions. The general chemical formula of the tourmaline-group minerals is $XY_3Z_6[T_6O_{18}](BO_3)_3V_3W$ (HENRY *et al.*, 2011). Our investigated samples, which were also characterised structurally, show a pronounced positive correlation between the $\langle Y-O \rangle$ distances and the (MnO + CuO) content in this site with $r^2 = 0.84$. There is no significant correlation between the Li content in the Y site and the $\langle Y-O \rangle$ distances. The valence states of Mn have not been determined, but we consider it mainly as Mn^{2+} . Another correlation, which is even better, shows a negative correlation between the $\langle Y-O \rangle$ distances and the Al_2O_3 content ($r^2 = 0.94$). In the structurally characterised samples the T site is only occupied by Si ($\langle T-O \rangle$ distances of ~1.618 Å). The samples at each locality generally show a strong negative correlation between the X-site vacancies and the MnO content ($r^2 \approx 0.9$ for all samples with $F < 0.9$ apfu). The Mn content in these tourmalines is dependent on the availability of Mn, at the formation temperature, as well as on stereochemical constraints. Based on various data ERTL

et al. (2012) suggest that increasing formation temperatures exist for tourmalines with increasing ($Fe^{2+} + Mn^{2+}$) contents. Our investigated samples also show evidence for a positive correlation between the Mn content (Fe content is only very low) and the formation temperature, because X-site vacancies decrease when the formation temperature increases (HENRY & DUTROW, 1996).

The very weak correlation between MnO and CuO ($r^2 = 0.01$) demonstrates that there is no evidence for a temperature dependent incorporation of Cu into the tourmaline structure. Hence, the Cu content in tourmaline is essentially dependent on the availability of Cu and on stereochemical constraints.

This work was funded by the Austrian Science Fund (FWF) project no. P23012-N19 to AE.

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